Physical and Chemical Changes



Physical Changes

The FORM of a substance changes, but not what it is made of

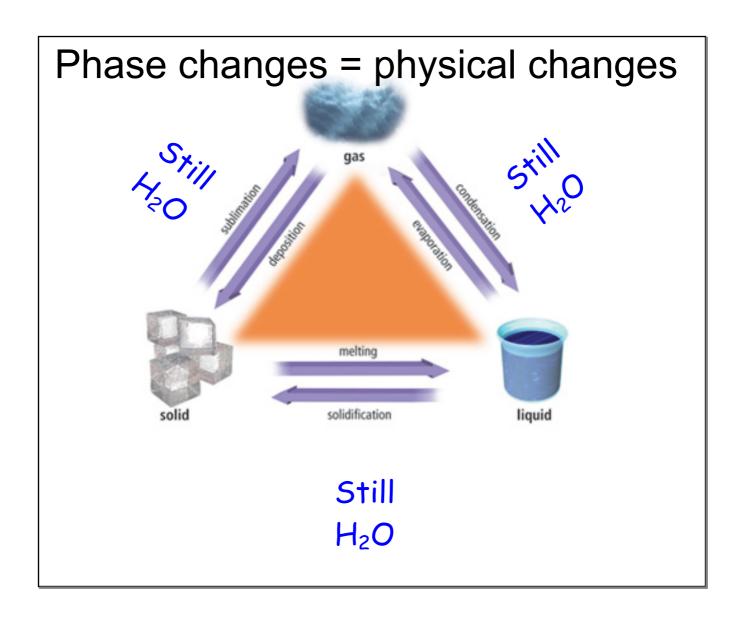




Physical Changes

Many physical changes can easily be reversed by changing temperature.





Chemical Changes

There is a change to arrangement of atoms in the compounds.

A new substance (or substances) are formed.

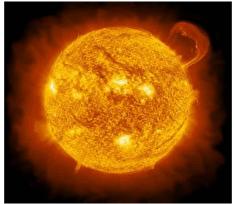
"Switching seating charts"
"Build new things out of the Legos"

Evidence that you have seen a chemical change:
Color changes

Evidence that you have seen a chemical change:

Energy is given off as heat or light



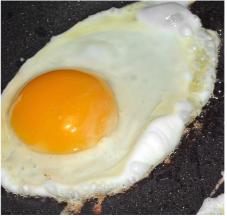




Evidence that you have seen a chemical change:

Energy is absorbed



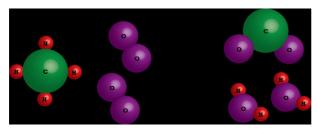


Evidence that you have seen a chemical change: The smell changes

Evidence that you have seen a chemical change:

New solids, liquids, or gases are

produced



 $6 CO_2 + 6 H_2O + sunlight --> C_6H_{12}O_6 + 6O_2$



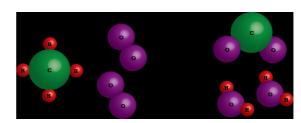
 $C_6H_{12}O_6 + 6O_2 --> 6 CO_2 + 6 H_2O + energy$

The Law of Conservation of Mass Matter can be neither created nor destroyed. It can only change form.



The Law of Conservation of Mass

In other words:

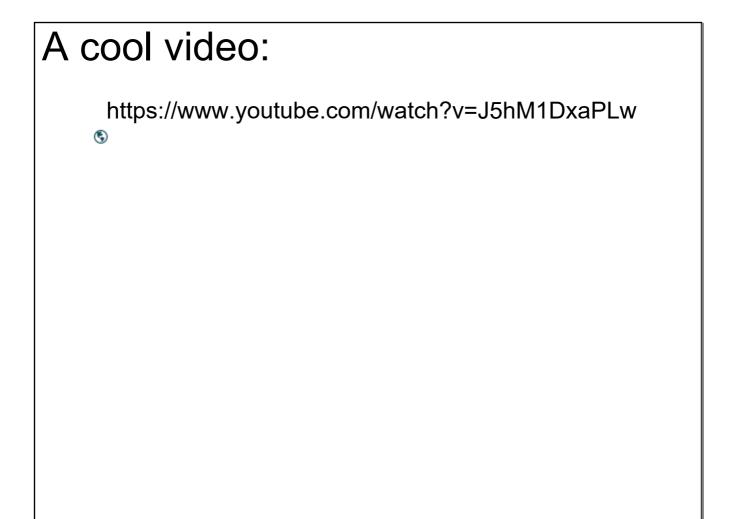


The mass of the reactants

will equal the mass of the products.

(mass before = mass after)

(sound familiar??)



Chemical or Physical Changes?:

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Video 1: Chemical or Physical Change?

Video 2: Chemical or Physical Change?

Video 3: Chemical or Physical Change

Video 4: Chemical or Physical Change?

Video 5: Chemical or Physical Change?
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